Vivekananda College of Engineering & Technology, Puttur

[A Unit of Vivekananda Vidyavardhaka Sangha Puttur @]
Affiliated to VTU, Belagavi & Approved by AICTE New Delhi

CRM08	Rev 1.10	<b.5.></b.5.>	<25/08/2021 >
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CONTINUOUS INTERNAL EVALUATION- 2

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Dept: B.S.	Sem / Div:II/A,B,C	Sub:Engineering Chemistry	S Code: 18CHE22			
Date:30/08/2021	Time: 3:00-4:30 PM	Max Marks: 50	Elective:N			
Note: Answer any 2 full questions, choosing one full question from each part.						

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()N	Questions	Marks	RBT	COs			
		PART A						
1		Define PV Cell. Explain principle, construction and working of PV	10	L1,L2	CO3			
	_	Cell. Mention their advantages and limitations						
	1	Define Knocking? Explain its mechanism with relevant chemical	8	L2	CO3			
	_	reaction.						
	c	Explain the preparation of solar grade silicon by union carbide	7	L2	CO3			
L		process.						
Ļ		OR						
2								
		ii) 0.85g of coal sample (Carbon 90%, H2 5% and ash 5%) was subjected						
		to combustion in a bomb calorimeter. Mass of water taken in the	10	1212	CO2			
		calorimeter was 2500g and the water equivalent of the calorimeter was 650g. The rise in temperature was found to be 3.2°C. Calculate the gross	10	L2,L3	CO3			
		and net calorific values of the sample.						
H	-	Define chemical fuel. Explain the experimental determination of	8	L1,L2	CO3			
		calorific value of fuel using Bomb calorimeter.	Ü					
		What are fuel cells. Explain the construction and working of solid	7	L1,L2	CO3			
		oxide fuel cell.						
PART B								
3	a	Define demineralisation. Explain ion exchange process for softening	10	L2	CO4			
		of water.						
	b	What is electroplating? Explain decorative electroplating chromium	8	L1,L2	CO2			
L		with relevant chemical reactions.						
	1	i) Define COD.						
		ii) In a COD experiment, 28.1ml and 14.0ml of 0.05N FAS solution	7	L1,L3	CO4			
		were required for blank and sample titration respectively. The volume						
L		of test sample used was 25ml. Calculate the COD of the sample?						
L.	OR							
[4		Explain the gravimetric and colorimetric method for estimation of	10	L1,L2	CO4			
-	_	sulphate and fluoride in water respectively.	0	1112	- CO2			
		What is electrolessplating? Explain electrolessplating of copper with	8	L1,L2	CO2			
H	-	relevant chemical reactions.	7	12	CO4			
	c	Define desalination. Describe the process of reverse osmosis of sea	/	L2	CO4			
L		water.						

Prepared by: Dr. Chethan P.D.

HOD: Prof. M. Ramananda Kamath

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